

## AMENDMENTS TO THE CLAIMS

The listing of claims will replace all prior versions, and listings, of claims in the application:

### Listing of Claims:

1. (Currently Amended) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:
  - an adsorption function metal oxide component comprising at least one metal oxide selected from the group consisting of the metals oxides of barium, cesium, lanthanum, strontium, and zirconium, and combinations thereof for said adsorption function metal oxide component providing adsorption function and adsorption sites for the nitrogen oxides in the fuel combustion gases; and
  - an oxidation function metal oxide component comprising at least one metal oxide selected from the transition metals group consisting of chromium, cobalt, copper, iron, manganese, and platinum, and combinations thereof for said oxidation function metal oxide component providing an oxidation function and sites closely adjacent to said adsorption metal oxide component and oxidation sites for the nitrogen oxides in the fuel combustion gases,
  - said adsorption metal oxide component and said oxidation metal oxide component being combined together in close intimate contact so that said adsorption metal oxide component and said oxidation metal oxide component are chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,
  - said adsorption sites and oxidation sites for nitrogen oxides within the bi-functional catalyst being adjacent to each other and arranged so as to promote adsorption and oxidation of nitrogen oxides when exposed to fuel combustion gases containing nitrogen oxides,
  - with the molar ratio of metal within the adsorption function metal oxide component metals to metal within the oxidation function metal oxide component metals being in a ratio range of 0.1:1 to 5:1 to provide in the bi-functional catalyst.

2. (Currently Amended) The bi-functional catalyst of claim 1, wherein said oxidation function metal oxide component includes copper, manganese, and platinum, and the molar ratio of metal within the said adsorption function metal oxide component metals to metal within the oxidation function metal oxide component metals is in a range of about 0.2:1 to 2:1.

3. (Currently Amended) The bi-functional catalyst of claim 1, wherein said adsorption function metal oxide component is includes at least one of barium (Ba) or lanthanum (La), and said metal oxide oxidation component is includes copper (Cu) and manganese (Mn) promoted with platinum (Pt).

4. (Currently Amended) The bi-functional catalyst of claim 2 1, wherein said adsorption metal oxide component is includes lanthanum (La), and said oxidation metal oxide components are component includes at least one of copper (Cu) and or manganese (Mn) promoted with platinum (Pt).

5. (Currently Amended) The bi-functional catalyst of claim 2 1, wherein said adsorption metal oxide component is includes barium (Ba), and said oxidation metal oxide component is includes copper (Cu), and optionally manganese (Mn), promoted with platinum (Pt).

6. (Currently Amended) The bi-functional catalyst of claim 1, wherein said bi-functional catalyst is adsorption and oxidation metal oxides in powder form are provided on a porous support material having surface area of at least about 50 m<sup>2</sup>/g.

7. (Currently Amended) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:

an adsorption function metal oxide component comprising at least one of barium or lanthanum ~~selected for and~~ providing adsorption function and adsorption sites for the nitrogen oxides in the fuel combustion gases; and

an oxidation function metal oxide component selected from the group consisting of copper, manganese and platinum ~~for and~~ providing oxidation function and sites closely adjacent to said adsorption metal oxide component and oxidation sites for the nitrogen oxides in the fuel combustion gases,

said adsorption metal oxides oxide component and said oxidation metal oxides oxide component being combined intimately together so that said adsorption metal oxide component and said oxidation metal oxide component are chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,

said adsorption sites and oxidation sites for nitrogen oxides within the bi-functional catalyst being adjacent to each other and arranged so as to promote adsorption and oxidation of nitrogen oxides when exposed to fuel combustion gases containing nitrogen oxides,

with the molar ratio of metal within the adsorption function metal oxide component metals to metal within the oxidation function metal oxide component metals being in a range of 0.2:1 to 2:1,

said metal oxides oxide components being provided on a porous inert support material having surface area of 50-500 m<sup>2</sup>/g ~~to provide the bi-functional catalyst.~~

8. (Currently Amended) A method for making a bi-functional catalyst suitable for catalytic oxidation of nitrogen oxides (NO<sub>x</sub>) contained in fuel combustion gases, comprising the steps of:

- (a) providing an aqueous solution of adsorption function metal ions selected from the metals group consisting of barium, cesium, lanthanum, strontium and zirconium and combinations thereof for providing an adsorption function and adsorption sites in the bi-functional catalyst for adsorbing nitrogen oxide contained in combustion gases;
- (b) providing an aqueous solution of oxidation function metal ions selected from the metals group consisting of chromium, cobalt, copper, iron, manganese and platinum for providing an oxidation function and oxidation sites for in the bi-functional catalyst oxidizing nitrogen oxides in combustion gases,  
said adsorption function metals metal ions having a molar ratio to the oxidation function metals metal ions in the range of 0.1:1 to 5:1;
- (c) mixing said adsorption and oxidation function metal ion solutions together and adding a binding agent acid having at least two functional groups selected for providing close intimate contact of said metal ions ~~and thereby providing to form~~ a precursor solution;
- (d) drying said metal ions precursor solution and heating to a temperature of 500-800°C (930-1470°F) ~~for 0.4-5 hours and thereby forming to form~~ a metal oxide complex precursor material, then cooling the metal oxide complex precursor material; and
- (e) forming a second solution of oxidation function metal ions and a platinum salt, and impregnating said precursor material with said second solution, ~~into~~, then drying and calcining the impregnated precursor material at 500-800°C temperature and cooling to provide yield the bi-functional catalyst.

9. (Currently Amended) The method for making a bi-functional catalyst of claim 8, wherein the binding agent acid is comprises carboxylic acid having a molar ratio of acid to total metals in a range of 0.5:1 to 2:1.

10. (Currently Amended) The method for making a bi-functional catalyst of claim 9, wherein the carboxylic binding acid is comprises citric acid.

11. (Currently Amended) The method for making a bi-functional catalyst of claim 9, wherein the carboxylic binding acid is comprises sodium citrate.

12. (Currently Amended) A method for making a bi-functional catalyst suitable for catalytic oxidation of nitrogen oxides (NOx) contained in fuel combustion gases, comprising the steps of:

(a) providing an aqueous solution of adsorption function metal ions selected from the metals group consisting of barium (Ba) and lanthanum (La) for providing an adsorption function and adsorption sites in the bi-functional catalyst for adsorbing nitrogen oxides contained in combustion gases;

(b) providing an aqueous solution of oxidation function metal ions selected from the metals group consisting of copper (Cu) and manganese (Mn) for providing an oxidation function and oxidation sites in the bi-functional catalyst for oxidizing nitrogen oxides contained in combustion gases,

said adsorption function metal ions metals having a molar ratio to the oxidation function metal ions metals in the range of 0.1:1 to 5:1;

(c) mixing said adsorption and oxidation function metal ion solutions together and adding a binding agent acid having at least two functional groups for providing close intimate contact of said metal ions and thereby providing to form a precursor solution;

(d) drying said metal ions precursor solution and heating to a temperature of 500-800°C (930-1470°F) ~~for 0.4-5 hours and thereby forming to form~~ a metal oxide complex precursor material, then cooling the metal oxide complex precursor material; and

(e) forming a second solution of oxidation function metal ions comprising manganese (Mn) ions and a platinum (Pt) salt, and impregnating said metal oxide complex precursor material with the second solution, then drying and calcining the impregnated precursor material at 500-800°C temperature and cooling to provide yield the bi-functional catalyst.

13. (Currently Amended) The method for making a bi-functional catalyst of claim 12, wherein said adsorption function metal ions comprise is barium (Ba).

14. (Currently Amended) The method for making a bi-functional catalyst of claim 12, wherein said adsorption function metal ions comprise is lanthanum (La).

15. (Currently Amended) The method for making a bi-functional catalyst of claim 12, wherein said adsorption function metal ions comprise is barium (Ba) and said oxidative function metal ions comprise is copper (Cu) and manganese (Mn) promoted with platinum (Pt).

16. (Original) The method for making a bi-functional catalyst of claim 8, including depositing the catalyst in powder form onto a porous support material.

17. (Currently Amended) A process for catalytic oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the process comprising:

(a) providing a combustion gas stream containing nitrogen oxides (NOx) ~~not exceeding about 0.5 vol%~~ together with ~~5-10 vol%~~ oxygen; and

(b) contacting said combustion gas stream with a the bi-functional oxidation catalyst ~~as defined by of~~ claim 1[[],] ~~in order for~~ said catalyst ~~being capable of adsorbing and oxidizing to adsorb and oxidize at least a portion of~~ the nitrogen oxides (NOx) in said combustion gas stream at a temperature in a range of 170-550°F, ~~temperature and a pressure in a range of~~ 0.5-20 psig, ~~pressure and a~~ space velocity in a range of 5,000-100,000 hr<sup>-1</sup> to form higher oxides of nitrogen and providing and thereby yield a treated combustion gas stream containing partially oxidized nitrogen oxides (NOx).

18. (Currently Amended) The catalytic oxidation process of claim 17, wherein said combustion gas stream contains up to 0.2 0.5 vol. % nitrogen oxides (NOx).

19. (Currently Amended) The catalytic oxidation process of claim 17, wherein the ~~catalytic oxidation conditions are~~ said combustion gas stream in (b) has a temperature in a range of ~~200-500°F~~ ~~220-350°F~~ temperature, a pressure in a range of 1-15 psig pressure, and a space velocity in a range of 8,000-50,000 h<sup>-1</sup>, and wherein the nitrogen oxides (NOx) concentration in the treated combustion gas stream is comprises at least about 50 vol % NO<sub>2</sub>.

20. (Currently Amended) The catalytic oxidation process of claim 19, wherein the nitrogen oxides (NOx) in the treated combustion gas stream contains 60-98 vol % NO<sub>2</sub>.

21. (Cancelled)

22. (Currently Amended) The catalytic oxidation process of claim 21 17, wherein said combustion gas stream contains up to 0.2 vol. % NO<sub>x</sub>.

23. (Currently Amended) The catalytic oxidation process of claim 21 17, further comprising:

(c) contacting said treated gas stream containing partially oxidized nitrogen oxides (NOx) with a chemical oxidant having a molar ratio of said chemical oxidant to nitrogen oxides (NOx) in a range of 0.5:1 to 1.2:1 to further oxidize the partially oxidized nitrogen oxides (NOx) to higher oxides of nitrogen; and

(d) scrubbing the treated combustion gas stream containing the higher oxides of nitrogen with a scrubbing liquid to remove the higher oxides of nitrogen from the treated combustion gas and yield a further treated combustion gas stream containing less than about 15 ppm nitrogen oxides (NOx)

~~wherein the catalytic oxidation conditions are 200-500°F temperature, 1.0-15 psig pressure, and space velocity of 8,000-50,000 h<sup>-1</sup>, and the NOx in the initially treated gas stream is at least about 50 vol % NO<sub>2</sub>.~~

24. (Currently Amended) The catalytic oxidation process of claim 21 23, wherein the ~~chemical oxidation reaction conditions are~~ (c) is performed at a temperature in a range of 100-250°F temperature and a pressure in a range of 0.8-1.4 psig pressure.

25. (Currently Amended) The catalytic oxidation process of claim 24 23, wherein said chemical oxidant is comprises ozone (O<sub>3</sub>), and wherein the molar ratio of the ozone (O<sub>3</sub>) to the nitrogen oxides (NOx) in said treated combustion gas stream NOx is 0.8:1 to 1.0:1.

26. (Currently Amended) The catalytic oxidation process of claim 24 23, wherein said scrubbing liquid is comprises water.

27. (Currently Amended) The catalytic oxidation process of claim 24 23, wherein said ~~oxidation catalyst is a bi functional oxidation catalyst as defined by claim 1 and the catalytic oxidation (b) is performed at a temperature is in a range of 300-350°F.~~

28. (Currently Amended) The catalytic oxidation process of claim 27 23, wherein the ~~final further treated combustion gas stream formed in (d) removed at step (e)~~ contains less than about 10 ppm nitrogen oxides (NOx).

29. (Cancelled)

30. (New) A bi-functional catalyst obtainable by the method of claim 8, wherein said bi-functional catalyst is suitable for catalytic oxidation of nitrogen oxides (NOx) contained in fuel combustion gases.

31. (New) A bi-functional catalyst obtainable by the method of claim 12, wherein said bi-functional catalyst is suitable for catalytic oxidation of nitrogen oxides (NOx) contained in fuel combustion gases.

32. (New) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:

an adsorption function metal oxide component comprising at least one metal oxide selected from the group consisting of oxides of barium, cesium, lanthanum, strontium, and zirconium and providing adsorption function and adsorption sites for nitrogen oxides in fuel combustion gases; and

an oxidation function metal oxide component comprising at least one metal oxide selected from the group consisting of chromium, cobalt, copper, iron, manganese, and platinum and providing an oxidation function and oxidation sites for nitrogen oxides in fuel combustion gases,

said adsorption metal oxide component and said oxidation metal oxide component being combined together in close intimate contact by means of an intermediate aqueous solution, comprising (i) at least type of adsorption function metal ions, (ii) at least one type of oxidation function metal ions, and (iii) a binding acid agent, that is heated and dried in order for said adsorption and oxidation metal oxide components to be chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,

the molar ratio of the adsorption function metal ions to the oxidation function metal ions in the intermediate aqueous solution being in a range of 0.1:1 to 5:1.

33. (New) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:

an adsorption function metal oxide component comprising at least one metal oxide selected from the group consisting of oxides of cesium, lanthanum, strontium, and zirconium; and

an oxidation function metal oxide component comprising at least one metal oxide selected from the group consisting of chromium, cobalt, copper, iron, manganese, and platinum,

said adsorption metal oxide component and said oxidation metal oxide component being combined together in close intimate contact so that said adsorption metal oxide component and said oxidation metal oxide component are chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,

the molar ratio of the metal within the adsorption function metal oxide component to the metal within the oxidation function metal oxide component being in a ratio range of 0.1:1 to 5:1 in the bi-functional catalyst.

34. (New) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:

an adsorption function metal oxide component comprising barium oxide and a substantial quantity of at least one other metal oxide selected from the group consisting of oxides of cesium, lanthanum, strontium, zirconium; and

an oxidation function metal oxide component comprising at least one metal oxide selected from the group consisting of chromium, cobalt, copper, iron, manganese, platinum,

said adsorption metal oxide component and said oxidation metal oxide component being combined together in close intimate contact so that said adsorption metal oxide component and said oxidation metal oxide component are chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,

the molar ratio of the metal within the adsorption function metal oxide component to the metal within the oxidation function metal oxide component being in a ratio range of 0.1:1 to 5:1 in the bi-functional catalyst.

35. (New) A bi-functional catalyst useful for oxidation and removal of nitrogen oxides (NOx) contained in fuel combustion gases, the catalyst comprising:

an adsorption function metal oxide component comprising at least one metal oxide selected from the group consisting of oxides of barium, cesium, lanthanum, strontium, and zirconium; and

an oxidation function metal oxide component comprising at least one metal oxide selected from the group consisting of chromium, cobalt, copper, iron, manganese, and platinum,

said adsorption metal oxide component and said oxidation metal oxide component being combined together in close intimate contact so that said adsorption metal oxide component and said oxidation metal oxide component are chemically bonded closely together so as to form a metal oxide complex having a crystalline structure,

the molar ratio of metal in the adsorption function metal oxide component to metal in the oxidation function metal oxide component being in a range of 1:3 to 1:6.